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Hybridization between K and C₆₀ Electronic States in Superconducting K₃Ba₃C₆₀

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We have studied the electronic structure of the superconducting fulleride K₃Ba₃C₆₀ using the local density approximation in the density functional theory. We have found the presence of the hybridization not only between Ba and C₆₀ states but also between K and C₆₀ states. This novel hybridization should have an effect on the electronic properties, including superconductivity. Also in Rb₃Ba₃C₆₀, the hybridization between Rb and C₆₀ is found as well. In addition, the density of states at Fermi level of K₃Ba₃C₆₀ is found to be slightly larger than that of Rb₃Ba₃C₆₀, although the lattice constant of K₃Ba₃C₆₀ is shorter than that of Rb₃Ba₃C₆₀. These results are in sharp contrast to A₃C₆₀ superconductors (A=alkali-metal atoms).

Keywords: Electronic structure; K₃Ba₃C₆₀; Rb₃Ba₃C₆₀; superconductivity; Density-Functional Theory

INTRODUCTION

The face-centered-cubic (fcc) A₃C₆₀ superconductors (A=K, Rb, etc.), have been studied intensively. Now it is well known that the larger the lattice constant is, the higher the superconducting transition temperature T_c is^[1], e.g., 29K in Rb₃C₆₀, ($a=14.384\text{\AA}$)^[2] versus 19K in K₃C₆₀, ($a=14.240\text{\AA}$)^[3]. Accordingly, Cs₂RbC₆₀, which has the largest lattice constant, has the highest T_c (33K) among fcc A₃C₆₀ superconductors^[4]. This monotonical relation between the lattice constant and T_c can be explained qualitatively from the density of states at Fermi level, $N(E_F)$, by the standard BCS-type theory based on their electronic structure^[5-9].

On the other hand, body-centered-cubic (bcc) A₃Ba₃C₆₀ superconductors (A=K, Rb, or their mixture) synthesized recently are found to show a different relationship between the lattice constant and T_c from A₃C₆₀^[10-12]. T_c

of $K_3Ba_3C_{60}$ is 5.6 which is higher than that of $Rb_3Ba_3C_{60}$ (2.0K), although the latter has the larger lattice constant than the former. In the case of similar alkaline-earth-doped bcc fullerides, Sr_6C_{60} and Ba_6C_{60} [13,14], it has been shown that C_{60} states are hybridized strongly with the alkaline-earth-metal d states [15,16]. the hybridization between C_{60} states and Ba states may play an important role in their electronic properties. In order to clarify the details of the hybridization, we have studied the electronic structure of $K_3Ba_3C_{60}$ and $Rb_3Ba_3C_{60}$ from the first-principles calculation.

COMPUTATIONAL METHOD

In $A_3Ba_3C_{60}$ to be studied, A and Ba atoms are experimentally reported to be randomly located in which all C_{60} clusters are orientationally ordered with the highest-symmetry orientation as in the case of K_6C_{60} and Ba_6C_{60} . In order to perform the electronic-structure calculation, we must assume the fixed location of A and Ba atoms, as shown in Fig. 1, which realizes the highest possible symmetry (C_3 point group) under the condition that the unit cell contains one C_{60} cluster. In $K_3Ba_3C_{60}$, atomic coordinates determined experimentally [10] are used. As for $Rb_3Ba_3C_{60}$ of which atomic coordinates have not been reported experimentally, we have optimized the geometry by using the conjugate-gradient procedure [23] under the reported lattice constant of 11.32 Å.

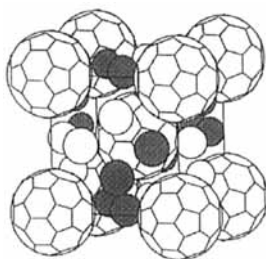


FIGURE 1 Structure of $K_3Ba_3C_{60}$ studied in the present work. Shaded and white spheres denote K and Ba atoms, respectively.

In the present electronic-structure calculations, we use the local density approximation (LDA) within the framework of the density functional theory [18,19]. We adopt the Ceperley-Alder exchange-correlation potential in the LDA [20]. The norm-conserving pseudopotentials [21] with the Kleinman-Bylander separable approximation [22] are also adopted. The real-space-partition method [23] is used in order to avoid the breakdown of the separable approximation for K, Rb, and Ba. A plane-wave basis set with a cutoff energy of 50 Ry is used.

RESULTS AND DISCUSSION

In Fig. 2, the band structure and density of states (DOS) of K₃Ba₃C₆₀ obtained are shown.

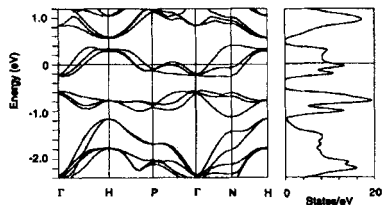


Figure 2 Band structure (left panel) and the density of states (right panel) of K₃Ba₃C₆₀. Energy is measured from the Fermi level denoted by the horizontal line. The band where the Fermi level lies is the t_{1g} -derived band. The density of states is broadened by using the Gaussian-distribution function with the width of 0.001eV.

The band in which the Fermi level lies is the t_{1g} -derived band (originating from the t_{1g} state, the second LUMO of the C₆₀ cluster) is half-filled with three electrons. The band structure of K₃Ba₃C₆₀ is, as will be discussed later, more dispersive than that of the pristine bcc C₆₀ having the same lattice constant as that of K₃Ba₃C₆₀. It is also the case in the band structure of Rb₃Ba₃C₆₀ which we have also calculated^[24]. This band broadening clearly indicates the presence of the hybridization. Surprisingly, not only the shapes but also t_{1g} -band widths of K₃Ba₃C₆₀ and Rb₃Ba₃C₆₀ are found to be quantitatively very similar to each other although the lattice constant of K₃Ba₃C₆₀ is smaller than that of Rb₃Ba₃C₆₀. Moreover, $N(E_F)$ of K₃Ba₃C₆₀ is 11.4 (states/eV), which is slightly higher than that of Rb₃Ba₃C₆₀, 11.2 (states/eV). From the viewpoint of the BCS theory, this is consistent with the experimental result that T_c of K₃Ba₃C₆₀ is higher than that of Rb₃Ba₃C₆₀.

These interesting result may be caused by the hybridization. Therefore, in order to study the hybridization behavior more quantitatively, we have calculated t_{1g} -band widths, which are found to be equivalent to their widths at the N point, for several hypothetical bcc fullerides with the fixed lattice constant of $a = 11.246 \text{ \AA}$ or 11.32 \AA . Systems studied are Ba₃C₆₀ where K (or Rb) atoms are removed from K₃Ba₃C₆₀ (Rb₃Ba₃C₆₀), and K₆C₆₀ with above a values. The atomic coordinates of K₆C₆₀ assumed are identical to those of K₃Ba₃C₆₀ for $a = 11.246 \text{ \AA}$, and to those optimized in Rb₃Ba₃C₆₀ for $a = 11.32 \text{ \AA}$.

Results are listed in Table I. These results have been obtained by treating K 3p and Ba 5p states as core states. We have also performed the

same calculation by treating them as valence states and found the following arguments hold irrespective of the treatment of K 3*p* and Ba 5*p* states^[24].

TABLE I Widths of *t*_{1*g*}-derived bands at the N point for several hypothetical bcc C₆₀ materials and the actual K₃Ba₃C₆₀.

	width at <i>a</i> = 11.246 Å	width at <i>a</i> = 11.32 Å
C ₆₀	0.34 eV	0.31 eV
K ₆ C ₆₀	0.56	0.54
Ba ₃ C ₆₀	0.56	0.57
A ₃ Ba ₃ C ₆₀	0.67 (A=K)	0.66 (A=Rb)

A considerable difference in *t*_{1*g*}-band widths of K₃Ba₃C₆₀ and bcc Ba₃C₆₀ at *a* = 11.246 Å indicates the presence of the hybridization between K and C₆₀ states in the K₃Ba₃C₆₀ superconductor. Also, the difference in *t*_{1*g*}-band widths of bcc Ba₃C₆₀ and bcc C₆₀ at *a* = 11.246 Å indicates the presence of the hybridization between Ba and C₆₀ states. The magnitude of these differences show that the degree of the hybridization between K and C₆₀ states is about half of that between Ba and C₆₀ states; this can be confirmed as well by the fact that the *t*_{1*g*}-band width of K₆C₆₀ is almost identical to that of Ba₃C₆₀ at *a* = 11.246 Å. Similarly, the difference in *t*_{1*g*}-band widths of Rb₃Ba₃C₆₀ and Ba₃C₆₀ at *a* = 11.32 Å has been obtained, indicating the presence of the hybridization between Rb and C₆₀ states.

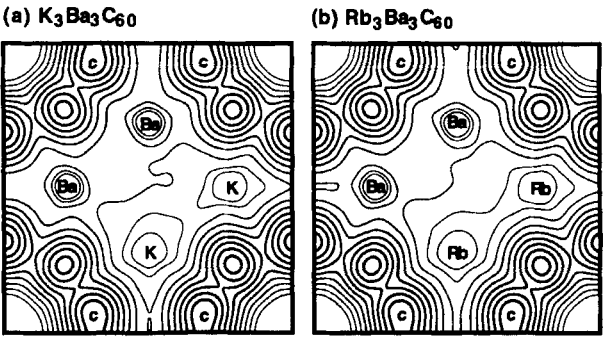


FIGURE 3 (a) Valence electron density on the (100) plane of K₃Ba₃C₆₀, (b) that of the pristine C₆₀. The center of the C₆₀ cluster is located at each corner. Each contour line indicates twice (half) the density of the neighboring thinner (thicker) contour lines. The highest-density contour lines correspond to 0.2 in atomic units.

This hybridization between alkali states and C₆₀ states is implied also by the valence-electron densities of K₃Ba₃C₆₀ and Rb₃Ba₃C₆₀ as shown in

Fig. 3. Figure 3 indicates that there remain considerable electrons around not only Ba but also alkali sites. Similarly, due to this hybridization, the superconducting carriers should appear not only on C₆₀ clusters and around Ba sites but also around alkali sites^[24]. Because the carriers should be coupled with the optical phonon involving alkali-ion displacements, the mass difference between K and Rb may cause the T_c difference.

The similar hybridization between C₆₀ between K states and C₆₀ states is found to be the present also in the actual K₆C₆₀ ($a = 11.39 \text{ \AA}$)^[25].

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